

## The Chemistry Kitchen

### Year 5 - Properties and Changes of Materials

### Key Knowledge

Different **materials** are used for particular jobs based on their properties: **electrical conductivity**, **flexibility**, **hardness**, **insulators**, **magnetism**, **solubility**, **thermal conductivity**, **transparency**. For example, glass is used for windows because it is hard and transparent.

### Sequence of learning:

Solids, liquids and gases  
 Properties of materials  
 Thermal conductivity  
 Reversible and irreversible changes  
 Burning  
 Dissolving  
 Separating mixtures

In this unit, children will compare and group materials based on their properties, know that some materials will dissolve in liquid to form a solution, learn to separate mixtures, demonstrate that dissolving, mixing and changes of state are reversible and be able to explain that some changes make new materials.

### Changes of State



### Key Skills

Begin to look at the accuracy, precision, repeatability and reproducibility of investigations.  
 With support, take measurements, using a range of scientific equipment, with increasing accuracy and precision, taking repeat readings when appropriate.  
 Begin to ask questions and develop a line of enquiry based on observations of the real world, alongside prior knowledge and experience.  
 With support, make predictions using scientific knowledge and understanding.

**What are the properties of materials?**

**Can we change them?**

### Key Vocabulary

<b>materials</b>	The substance that something is made out of, e.g. wood, plastic, metal.
<b>solids</b>	One of the three states of matter. <b>Solid</b> particles are very close together, meaning <b>solids</b> , such as wood and glass, hold their shape.
<b>liquids</b>	This state of matter can flow and take the shape of the container because the particles are more loosely packed than solids and can move around each other. Examples of <b>liquids</b> include water and milk.
<b>gases</b>	One of the three states of matter. <b>Gas</b> particles are further apart than <b>solid</b> or <b>liquid</b> particles and they are free to move around. Examples of <b>gases</b> are oxygen and helium.
<b>melting</b>	The process of heating a <b>solid</b> until it changes into a <b>liquid</b> .
<b>freezing</b>	When a <b>liquid</b> cools and turns into a <b>solid</b> .
<b>evaporating</b>	When a <b>liquid</b> turns into a <b>gas</b> or vapour.
<b>condensing</b>	When a <b>gas</b> , such as water vapour, cools and turns into a <b>liquid</b> .